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a state of balance in which the rates of the forward and reverse reactions are equal; no net change in the amount of reactants and products occurs in the chemical system (18.2) equilibrium position the relative concentrations of reactants and products of a reaction that has reached equilibrium; indicates whether the reactants or products are favored in the reversible reaction (18.2)

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Chapter 18 Reaction Rates and Equilibrium. Flashcard maker : Joseph Fraser. How is the rate of a chemical change expressed? in chemistry, the rate of chemical change or the reaction rate is usually expressed as the amount of reactant changing per unit time.

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Chapter 18 - Reaction Rates and Equilibrium - 18.3 ...
Chapter 18 Reaction Rates And Equilibrium. In layman’s terms, equilibrium is defined as a state of balance due to equal reactions of opposing forces, and today we’ll be talking all about it with regards to the scientific study of chemistry, focusing on such topics as reaction rates.

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Chapter 18 Reaction Rates and equilibrium. STUDY. PLAY. rate. is a measure of how much something changes within a specified amount of time. reactant, product. In chemistry, the rate of a chemical reaction, or the reaction rate is usually expressed as the change in the amount of _____ or _____ per unit of time. ...

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Chapter 18: Reaction Rates and Equilibrium. STUDY. PLAY. chemical kinetics. study of reaction rates and the factors that affect the rates. collision theory. for a reaction to occur, molecules/atoms must first collide-only some collisions are effective and lead to a reaction

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Chapter 18 Reaction Rates and Equilibrium193. SECTION 18.1 RATES OF REACTION (pages 541–547) This section explains what is meant by the rate of a chemical reaction. It also uses collision theory to show how the rate of a chemical reaction is influenced by the reaction conditions. Collision Theory (pages 541–544)

Name Date Class REACTION RATES AND EQUILIBRIUM 18
Chapter 18 Reaction Rates and Equilibrium. Description. Key Concepts and Vocabulary. Total Cards. 39. Subject. Chemistry. ... How is the rate of a chemical change expressed? Definition. in chemistry, the rate of chemical change or the reaction rate is usually expressed as the amount of reactant changing per unit time. Term. What four factors ...

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Figure 18.2, page 542: compare the rates A “rate” is a measure of the speed of any change that occurs within an interval of time In chemistry, reaction rate is expressed as the amount of reactant changing per unit time. Example: 3 moles/year, or 5 grams/second

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Chapter 18 Review “Reaction Rates and Equilibrium” Name: _____ 1. Energy that is available to do work is called free energy. 2. Reaction rate is defined as the number of atoms, ions, or molecules that react in a given time to form products. 3.

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Reversible Reactions If the rate of the shoppers going up the escalator is equal to the rate of the shoppers going down, then the number of shoppers on each floor remains constant, and there is an equilibrium. 18.2 9