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First we need the
mass flow and
velocity from each
hole "i," $i = 1$ to 6 : i

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ii2 i Q (3.07448.8)/6
ft Q 3/448.8 slug V
5.81 m 1.94
0.00216 As66s3/16

(/4) 12 Recall

Example 3.15 from
the text. For each
hole, we need the
absolute velocity,
 $V_i r_i$.

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Solution: First we
need the mass flow
and velocity from
each hole "i," $i = 1$ to
6: $\dot{m}_i = \rho V_i A_i$

$(3.0/448.8)/6 \text{ ft}^3/\text{s}$

$3/448.8 \text{ slug/s}$ $V = 5.81$

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m 1.94 0.00216
As66s3/16 (/4) 12
Recall Example
3.15 from the text.
For each hole, we
need the absolute
velocity, V_i r_i . 256
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Solution: First we

need the mass flow

and velocity from

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each hole "i," i 1 to
6: i ii2 i Q
(3.0/448.8)/6 ft Q
3/448.8 slug V 5.81
m 1.94 0.00216
As66s3/16 (/4) 12
Recall Example
3.15 from the text.

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Solution: First we
need the mass flow
and velocity from
each hole "i," $i = 1$ to
 6 : $i \quad ii \quad 2 \quad i \quad Q$
 $(3.0/448.8)/6 \text{ ft } Q$
 $3/448.8 \text{ slug } V \quad 5.81$
 $m \quad 1.94 \quad 0.00216$
 $As \quad 66s \quad 3/16 \quad (/4) \quad 12$

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Recall Example
3.15 from the text.
For each hole, we
need the absolute
velocity, V_i .

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Fluid Mechanics,

HW#3 2.56

Determine the

pressure of the

water in pipe A

shown in Fig. P2.56

if the gage

pressure of the air

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in the tank is 14
kPa. $P = 14$ kPa Air
 $SG = 0.9$ 0.3 m 1.2
m 0.6 m - Water
Figure P2.56

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Mechanics, HW#3
2.56 Determine
The Pressure ...
Chapter 3 Pressure
and Fluid Statics
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CHAPTER 3

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Then the
manometer rule
gives $p_A + \rho g h_{AB}$
 $+ \rho g h_{BC} -$

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$rH_ggh_{CD} = p_D$

Here, $p_A = p_D = 0$,
since points A and
D are exposed to

the atmosphere. 0

+ 1900 kg > m³

21g210.6 m² +

11000 kg > m³

21g210.8 m - h ...

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Fluid Mechanics,
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1.3. The Maxwell probability distribution, $f(v) = f(v_1, v_2, v_3)$, of molecular velocities in a gas flow at a point in space with average velocity u is given by (1.1). a) Verify

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that u is the
average molecular
velocity, and
determine the
standard
deviations (σ_1 ,

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Solution: $p_{abs} =$
140 kPa 1 14:70
psi 101:3 kPa =
20:32 psia p_{gage}

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$$= p_{\text{abs}} - p_{\text{atm}} = (20.32 \text{ psia}) - (14.70 \text{ psia}) = 5.62 \text{ psi}$$

gage = 5.62 psig

c.) Situation:

Pressure values
need to be

converted. Find:

Calculate the

absolute pressure
(psia)

corresponding to a
pressure of 0.55
bar (gage).

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Properties: $p_{\text{atm}} = 14.70 \text{ psi}$; Solution:
 $p_{\text{gage}} = 0.55 \text{ bar}$
1 ...

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