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ROUND 2 Exercise 7—2: Predict and balance the following synthesis and decomposition reactions. Use abbreviations to 1. 2. 3. 4. 5. 6. 8. 9. 10.

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The Ultimate Chemical Equations Handbook Determine the oxidation number of the underlined element: KMnO_4 Since K is an alkali metal, its charge must be 1+ Oxygen is 2— but there are four of them, therefore, 4 times 2— equals 8— If 1+ and 8— are added Ultimate Chemical Equation Answers

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