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#2 Biochemistry Lecture (Acids/Bases)

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## Acids and Bases - BIOLOGY JUNCTION

Acids and Bases . The degree of .  
acidity. or . alkalinity (basic) is  
important in organisms. The body  
must constantly maintain a near  
neutral pH (7) in the blood and body

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tissues. To do this, the body produces . buffers . that can . neutralize. acids. Acidic and basic conditions in the body occur due to different

## Biology Tests and Procedures |

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Nucleic Acids - BIOLOGY JUNCTION

A special property of acids and bases is their ability to neutralize the other's properties. In an acid-base (or neutralization) reaction, the H + ions from the acid and the OH-ions from the base react to create water (H 2 O).

## Acids And Bases Biology Junction

### Answers

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Answers Acids and Bases in

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important in living things because most enzymes can do their job only at a certain level of acidity. Cells secrete acids and bases to maintain the proper pH for enzymes to work.

## Acids And Bases Biology Junction Answers

Acids and Bases in Organisms Acids and bases are important in living things because most enzymes can do their job only at a certain level of acidity. Cells secrete acids and bases to maintain the proper pH for enzymes to work. For example, every time you digest food, acids and bases are at work in your digestive system.

## 1.20: Acids and Bases in Biology - Biology LibreTexts

According to the Lowry-Bronsted definition, an acid is a proton donor

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and a base is a proton acceptor.

According to the Lewis definition, acids are molecules or ions capable of coordinating with unshared electron pairs, and bases are molecules or ions having unshared electron pairs available for sharing with acids.

## Acids and Bases - Definition, Examples, Properties, Uses ...

There are several methods of defining acids and bases. While these definitions don't contradict each other, they do vary in how inclusive they are. The most common definitions of acids and bases are Arrhenius acids and bases, Brønsted-Lowry acids and bases, and Lewis acids and bases. Antoine Lavoisier, Humphry Davy, and Justus Liebig also made observations regarding acids and bases, but didn't formalize definitions.

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## Acids and Bases Terms and Definitions - ThoughtCo

Three consecutive bases on DNA called a triplet (e.g. TCG, ATG, ATT) mRNA codon table tells what 3 bases on mRNA code for each amino acid (64 combinations of 3 bases)

Methionine (AUG) on mRNA is called the start codon because it triggers the linking of amino acids; UAA, UGA, & UAG on mRNA signal ribosomes to stop linking amino acids together

## Nucleic Acids & Protein Synthesis - BIOLOGY JUNCTION

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on the above skills and collect up to  
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Biology: Acids/bases. STUDY.  
Flashcards. Learn. Write. Spell. Test.  
PLAY. Match. Gravity. Created by.  
kristinabrown6. Terms in this set (35)  
Acid. Any chemical that donates a  
hydrogen ion, proton donor. Base. Any  
chemical that accepts a hydrogen ion,  
proton acceptor. We say that a  
substance \_\_\_\_\_ as an acid or base.

Biology: Acids/bases Flashcards |  
Quizlet

Acids and Bases in Organisms Acids  
and bases are important in living  
things because most enzymes can do  
their job only at a certain level of  
acidity. Cells secrete acids and bases

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to maintain the proper pH for enzymes to work. For example, every time you digest food, acids and bases are at work in your digestive system.

## Acids and Bases ( Read ) | Biology | CK-12 Foundation

Cells secrete acids and bases to maintain the proper pH for enzymes to do their work. Every time you digest food, acids and bases are at work in your digestive system. Consider the enzyme pepsin, which helps break down proteins in the stomach. Pepsin needs an acidic environment to do its job.

## 3.12: Acids and Bases - Biology LibreTexts

Acids and Bases: Pepsin in Biology  
CK-12 PLIX Questions Name: Landen  
White Answer the following questions

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before you start using the simulation.

\*\*\*Use complete sentences 1. What are enzymes and what are they used for?\*\*\* Enzymes are biological molecules used by living systems to catalyze specific reactions.

Landen White - Acids and Bases Pepsin in Biology CK-12 ...

Nucleic Acids - BIOLOGY JUNCTION

According to the Lowry-Bronsted definition, an acid is a proton donor and a base is a proton acceptor.

According to the Lewis definition, acids are molecules or ions capable of coordinating with unshared electron pairs, and bases are molecules or ions having unshared electron pairs available for sharing with acids.

Acids And Bases Biology Junction  
Answers

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Acids and bases are important in the human body. For example, the stomach secretes hydrochloric acid, HCl, to digest food. The pancreas secretes a fluid rich in the base bicarbonate to neutralize stomach acid before it reaches the small intestine. Acids and bases react with metals.

## 10 Facts About Acids and Bases - ThoughtCo

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## pH Scale: Acids, bases, pH and buffers (article) | Khan ...

Acids and Bases in Organisms Acids and bases are important in living things because most enzymes can do

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their job only at a certain level of acidity. Cells secrete acids and bases to maintain the proper pH for enzymes to work. For example, every time you digest food, acids and bases are at work in your digestive system.

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