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## Pogil Answers Acids And Bases

An Arrhenius acid is any substance that produces hydrogen ions [or hydronium ions ( $\text{H}_3\text{O}^+$ ) a hydrogen ion attached to a water molecule] when dissolved in water. An Arrhenius base is any substance that produces

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hydroxide ions when dissolved in water. While the Arrhenius definitions of acids and bases is useful, it is limited.

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reactions. occur.

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STOP Conjugate Acid  $\text{CO}_2(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{HCO}_3^-(\text{aq}) + \text{H}^+(\text{aq})$   
Model 3 — Conjugate  
Acid—Base Pairs Base + 1-120 (l)  
 $\text{HCO}_3^-(\text{aq})$  13. All acid—base

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reactions have two conjugate acid—base pairs. One conjugate acid—base pair in the reaction in Model 3 is  $\text{H}_3\text{O}^+/\text{H}_2\text{O}$ . List the other acid—base pair in the reaction. 14.

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when equal molarities of acid and

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conjugate base are combined? 15.  
When unequal molarities of acid and conjugate base are combined to make a buffer (as in beaker 4 of Model 2), does the pH change significantly from the  $pK_a$  of the weak acid? Support your answer with data from Model 2.  
16. An industrial process requires a

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Phosphoric acid ( $\text{H}_3\text{PO}_4$ ) is a triprotic acid, but phosphorous acid ( $\text{H}_3\text{PO}_3$ ) is a diprotic acid, and

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hypophosphorous acid ( $\text{H}_3\text{PO}_2$ ) is a monoprotic acid. a. Draw Lewis dot-diagrams for each of these three acids.

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b.  $\text{HNO}_2$  is a weak acid, but  $\text{HNO}_3$  is a strong acid  $\text{HNO}_2$  has one terminal oxygen, but  $\text{HNO}_3$  has two, making it a stronger acid. c.  $\text{H}_3\text{PO}_4$  is a stronger acid than  $\text{H}_3\text{AsO}_4$  P is more electronegative than As, making

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H<sub>3</sub>PO<sub>4</sub> a stronger acid. d. FCH<sub>2</sub>CO<sub>2</sub>H  
is a stronger acid than CH<sub>3</sub>CO<sub>2</sub>H  
Electronegative fluorine withdraws  
electron density from the -OH bond of  
the carboxyl

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10 - Solutions Weak Acid ...

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a double-stranded, helical nucleic acid molecule capable of replicating and determining the inherited traits of organisms and viruses. Nucleotide a building block of DNA or RNA, consisting of a sugar, a nitrogenous base and a phosphate group. Double helix a pair of parallel helices

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intertwined about a common axis  
Hydrogen bond

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