

## Alkaline Pyrogallol Solution

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The aquatic plant *Myriophyllum spicatum* produces pyrogallic acid. When in alkaline solution, it absorbs oxygen from the air, turning brown from a colourless solution. It can be used in this way to calculate the amount of oxygen in air, notably via the use of the Orsat apparatus .

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~~Pyrogallol—Wikipedia~~

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Pyrogallol undergoes quick autooxidation in alkaline solution. Due to this property, its alkaline solution is generally used for the removal of oxygen from gases. Pyrogallol may also be used to prepare pyrogallol monomethyl ether via methylation and gallacetophenone via a modified Nencki reaction.

~~Pyrogallol ACS reagent | 87-66-1 | Sigma-Aldrich~~

Pyrogallol or benzene-1,2,3-triol is a white crystalline powder and a powerful reducing agent. It was first prepared by Scheele 1786 by heating gallic acid. An alternate preparation is heating...

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Pyrogallol or benzene-1,2,3-triol is a white crystalline powder and a powerful reducing agent. It

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was first prepared by Scheele 1786 by heating gallic acid. An alternate preparation is heating para-chlorophenoldisulphonic acid with potassium hydroxide. When in alkaline solution, it absorbs water readily, turning purple from a colourless solution.

~~what is alkaline pyrogallol solution? | Yahoo Answers~~

Pyrogallol reacts with dioxygen in weakly alkaline solutions to form several intermediate products which are electroactive substance and can be detected by electroanalytical methods. The focus here was putted on the effect of pH on the autoxidation rate of some intermediate products produced in pyrogallol autoxidation, which gives sensitive second-order derivative cathodic waves at  $\approx 0.20$ ,  $\approx 0.96$  and  $\approx 1.45$  V vs. SCE, respectively.

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50mg of pyrogallol was dissolved in 8ml of water and 1ml of 6% NaOH is then added. The pyrogallol is oxidized to a purple colour but also bizarre orange and ...

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Pyrogallol is the oldest photographic developer, its rapid deposition of silver from silver salts having first been noted in 1832. Alkaline solutions of pyrogallol absorb oxygen efficiently and are used in determining the oxygen content of gas mixtures.

~~Pyrogallol | chemical compound | Britannica~~

Alkaline Pyrogallol Solution Alkaline Pyrogallol - from Vickers Laboratories Limited To guard against the presence of oxygen in the nitrogen, an oxygen scrubbing solution such as an alkaline pyrogallol trap may be used. Prepare pyrogallol trap as follows: 1. Add 4.5 g pyrogallol

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Dissolve 20 g of resublimed pyrogallol in water, add 10 ml of conc. HCl and 2 g of  $\text{SnCl}_2 \cdot 2\text{H}_2\text{O}$  (dissolved in 5 ml of conc. HCl), and dilute the solution with 0.1 M HCl to 100 ml. Standard tantalum solution: 1 mg/ml. Fuse 0.1220 g of  $\text{Ta}_2\text{O}_5$ , with 4 g of  $\text{K}_2\text{S}_2\text{O}_7$  in a silica or platinum crucible. Dissolve the melt in 4%  $(\text{NH}_4)_2\text{C}_2\text{O}_4$  solution and dilute to the mark with the same reagent in a 100-ml standard flask.

~~Pyrogallol—an overview | ScienceDirect Topics~~

Procedure--The pyrogallol solution was placed in a 1 liter beaker and mixed with an electrically driven glass stirrer at room temperature. 5 ml. of hydrogen peroxide solution and 5 ml. of catalase solution were added. Stirring was continued for 1 minute. The remaining 115 ml.

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