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Neutralization reactions are one type of chemical reaction that proceeds even if one reactant is not in the aqueous phase. For example, the chemical reaction between $\text{HCl}(\text{aq})$ and $\text{Fe}(\text{OH})_3(\text{s})$ still proceeds according to the equation. $3 \text{HCl}(\text{aq}) + \text{Fe}(\text{OH})_3(\text{s}) \rightarrow 3 \text{H}_2\text{O}(\text{l}) + \text{FeCl}_3(\text{aq})$

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even though $\text{Fe}(\text{OH})_3$ is not soluble.

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Reactions Chapter Answer. H_2SO_4
(aq) + $\text{Sr}(\text{OH})_2$ (aq) \rightarrow $2\text{H}_2\text{O}$ (l) +

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~~Chapter 21 1~~
SrSO₄ (aq) Neutralization reactions are one type of chemical reaction that proceeds even if one reactant is not in the aqueous phase.

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A neutralization reaction is a reaction

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in which an acid and a base react in an aqueous solution to produce a salt and water. The aqueous sodium chloride that is produced in the reaction is called a salt. A salt is an ionic compound composed of a cation from a base and an anion from an acid.

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~~4.4: Neutralization Reactions - Chemistry LibreTexts~~

Read Chapter 19, pages 617 - 621 in the Glencoe - Chemistry Matter & Change textbook. Read Chapter 4, pages 149 - 154 (Section 4.8) in the Zumdahl - Chemistry textbook.

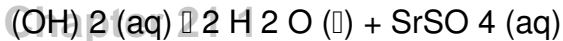
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Complete the "Neutralization Reactions: Titrations" booklet. (File and answer key below)

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Answer Key Neutralization Reactions
Chapter Answer. $\text{H}_2\text{SO}_4(\text{aq}) + \text{Sr}$

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ionic equation for the following
neutralization reactions (a) HI

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Neutralization Reactions Worksheet
Answers can be created by making use of different equations and calculating data. The task of creating the worksheet is actually very simple. All you need to do is to use a spreadsheet program and make the

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necessary adjustments to create the
neutralization reaction chart.

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1 chemical changes answer key. Many
chemical reactions necessary for life
happen at temperatures that would
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even if one reactant is not in the

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aqueous phase. For example, the chemical reaction between $\text{HCl}(\text{aq})$ and $\text{Fe}(\text{OH})_3(\text{s})$ still proceeds according to the equation. $3\text{HCl}(\text{aq}) + \text{Fe}(\text{OH})_3(\text{s}) \rightarrow 3\text{H}_2\text{O}(\text{l}) + \text{FeCl}_3(\text{aq})$ even though $\text{Fe}(\text{OH})_3$ is not soluble.

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gauge your knowledge of neutralization reactions and their characteristics. You will be assessed on acidic and basic substances.

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Combustion. Neutralization. Solvent.

Decomposition. a chemical reaction

that occurs between a fuel and an

oxidizing agent. a chemical reaction in which

an acid and a ...

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Key Concepts: Terms in this set (22)
Chemical reaction. ... the product of a
neutralization reaction in addition to
water. THIS SET IS OFTEN IN
FOLDERS WITH... Chapter 5 vocab,

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Corwin. 21 terms. ... Chapter 6

Chemical Reaction. 19 terms.

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Reactions. 31 terms.

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