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How To Calculate The Equilibrium

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Chemistry: 9.1-9.3, 9.5, 7.14 Entropy and

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Equilibrium Study Guide

- Write the equilibrium expression. Look up the value for  $K_a$  (in a table).
- Write down the initial and equilibrium

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concentrations for everything except pure water. • We usually assume that the equilibrium concentration of  $H^+$  is  $x$ .

- Substitute into the equilibrium constant expression and solve.
- Remember to convert  $x$  to pH if necessary.

AP Chemistry— CHAPTER 16 STUDY

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GUIDE Acid-Base Equilibrium

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AP Chemistry – Chapter 13, Chemical  
Equilibrium Study Guide & Ch. 16a  $K_{sp}$ .

Students should be able to... Describe a .  
system in equilibrium . Find the reaction .  
quotient,  $Q$ , ( $K_{eq}$ ) for a reaction equation .

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Solve for Q given . initial concentrations or pressures. Using . LeChatlier's principle

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chemical bonding, phases of matter,  
solutions, types of reactions, equilibrium,

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reaction kinetics, and thermodynamics.  
Anyone wishing to take an AP Chemistry course should take at least two years of high school math before enrolling in this course.

AP Chemistry Help, Notes, Outlines and Equations ...

Energy plays a huge role in AP Chemistry

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and in chemistry as a whole. When chemical reactions occur, a huge part of what happens has to do with energy entering or exiting a system, usually through heat. You'll learn about calorimetry, enthalpy, and bond energy and understand the role of energy in chemistry.

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Guides 28 Using the Unit Guides 31 UNIT  
1: Atomic Structure and Properties

AP Chemistry Course and Exam  
Description, Effective Fall 2020

The Chemistry of Acids & Bases 4

RELATIVE STRENGTHS OF ACIDS

AND BASES Strength is determined by the



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position of the "dissociation" equilibrium.

$\frac{3}{4}$  Strong acids/strong bases 1. dissociates completely in water 2. have very large dissociation or K values  $\frac{3}{4}$  Weak

acids/weak bases 1. dissociate only to a slight extent in water 2.

AP\*Chemistry The Chemistry of Acids and

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Bases

AP® /College Chemistry. Unit: Chemical equilibrium. Lessons. Equilibrium constant. Learn. Reactions in equilibrium (Opens a modal) ... (Opens a modal) Keq derivation intuition (Opens a modal) Heterogeneous equilibrium (Opens a modal) Calculating equilibrium constant  $K_p$  using partial

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The Best AP Chemistry Notes to Study With  
Description from the College Board Website  
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the equivalent of the general chemistry course usually taken during the first college year. The goal is that students will take the AP Exam to receive college credit or placement at the student ' s college of choice.

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