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Electrons Atomic structure

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blanks for the elements in
this chart. For the purposes
of this chart, round all
atomic masses to the nearest
whole number. Element Number
of Protons Number of

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Atomic Mass Atomic Number.
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Isotopes are atoms of the same element with different numbers of neutrons which results in different mass numbers.

Chemistry A Study Of Matter
3.6 Answer Key

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Atomic Structure Answers 1.

In the nuclear model of the atom, protons (and neutrons) are housed in a small, dense nucleus. Electrons surround the nucleus in an area of mostly empty space. 2. If electrons are electrically attracted to nucleus and would, therefore, have centripetal acceleration in order to orbit the nucleus.

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Atomic Structure: Chapter
Problems

Question 6. Define the
following terms: (a) Atomic
number (b) Mass number (c)
Nucleons (d) Valence shell

Answer: (a) Atomic number:
Atomic number refers to the
number of protons present in
an atom. It is denoted by Z .
Example: An atom of oxygen
contains 8 proton Therefore
its atomic number is 8.

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Atomic Structure In-Class

Worksheet Name Datu

Directions: read and answer

each question in order to

receive credit (1) Indicate

whether each of the

following is a permissible

set of quantum numbers. If

the set is not permissible,

state why it is not a.

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Answers Chemistry
2, 4, 0, $m = +1$ b. $n = 5, 3, m = -3$ c. $n = 4, l = 3, m = 2$ d. $n = 0, l = 0, m = 4, l = 1$
(2) Indicate whether each of the following is a permissible set of quantum numbers.

Solved: CHM 1045 General
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