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QUESTION 1 SOLUTION AND EXPLANATIONS (a) A dative covalent or coordinate covalent bond is one in which one of the atoms supplies both the shared electrons to the covalent bond. A dative covalent bond is written A → B, where the direction of the arrow shows the direction in which the electron pair is donated.

CAPE Chemstry 2012 U1 P2 Q1 Past Paper Answer And Explanations

Main Points To Be Used In Answering A Past Paper Question: Down the group there is a change in structure from giant molecular to giant metallic and a change in bonding from covalent to metallic. From C to Ge elements exhibit a giant molecular structure. Sn and Pb exhibit a giant metallic lattice structure.

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the bond is broken. (ii) The bond lengths and bond energies of carbon-carbon single, double and triple bonds are listed below.

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