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Chapter 10 Supplemental Problems Balance the following chemical equations. 1)  $\text{SnS}_2 (\text{s}) + \text{O}_2 (\text{g}) \rightarrow \text{SnO}_2 (\text{s}) + \text{SO}_2 (\text{g})$

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2 (g) 2)  $C_2H_6(g) + O_2(g) \rightarrow CO_2(g) + H_2O(g)$  3)  $Al(s) + HCl(aq) \rightarrow AlCl_3(aq) + H_2(g)$  4)  $CoCO_3(s) \rightarrow CoO(s) + CO_2(g)$  Write a balanced equation for each of the following reactions, substituting symbols and formulas for names.

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