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Chapter 12 Solutions review. liquid that dissolves a gas or solid or the solution component in greater amount. when a system in equilibrium is altered by a change of temperature, pressure, or concentration variable, the system shifts in equilibrium composition in a way that tends to counteract this change of variable.

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Chapter 12 Review: Solutions Name: Carly 1) Does a solution have to involve a liquid? Explain your answer. No; solutions may exist as gases, liquids, or solids. 2) Explain why a suspension is considered a heterogeneous mixture. The particles are so large that they settle out unless the mixture is constantly stirred or agitated.

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b colloids (c) smaller than 1 nm 2. Identify the solvent in each of the following examples:

mc06se cFMsr i-vi

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Chapter 12, Problem Review_and_Discussion 13 : 5. How does ...

Modern Chemistry Chapter 12 Solutions Review Answers Modern Chemistry Chapter 12-Solutions Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Chapter 12 Review Solutions Section 3 Answers

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