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~~Answer Key Chapter 12: Stoichiometry Mole Ratios Questions 1. Aluminum reacts with oxygen to produce aluminum oxide as follows: 4Al + 3O₂ → 2Al₂O₃ a. If you use 2.3 moles of Al, how many moles of Al₂O₃ can you make? b. If you want 3.9 moles of Al₂O₃, how many moles of O₂ are needed? 2.~~

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Merely said, the chapter 9 stoichiometry work answers is universally compatible past any devices to read. chapter 9 stoichiometry work answers CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: C₃H₄(g) + xO₂(g) → 3CO₂(g) + 2H₂O(g) 4 a.

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