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To solve quantitative problems involving the stoichiometry of reactions in solution. Quantitative calculations involving reactions in solution are carried out in the same manner as we discussed in Chapter 11 .

Chapter 12.2: Stoichiometry of Reactions in Solution ...

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Stoichiometry is the calculation of the amount of substances in a chemical reaction from the balanced equation. The sample problem below is another stoichiometry problem involving ingredients of the ideal ham sandwich. Example 12.1. 1 Kim looks in the refrigerator and finds that she has 8 slices of ham.

12.1: Everyday Stoichiometry - Chemistry
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Actual & Percent yield Part I.

Stoichiometry 1. $1 \text{ ZnI}_2 \rightarrow 1 \text{ Zn} + 1 \text{ I}_2$

How many grams of iodine will you produce if you begin your rxn with 56.7g of zinc iodide? 56.7 g 1 mole 1 mole 253.8 g 319.19 g 1 mole 1

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Answer Key

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