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~~Boyle's law. Charles's law. a temperature at which a gas~~

~~theoretically occupies zero volume. 0.08206 L atm/mol K. at~~

~~constant temperature, the volume of a given amount of gas is \propto .~~

~~at constant pressure, the volume of a given amount of gas is \propto .~~

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temperature at which a gas theoretically occupies zero volume.
0.08206 L atm/mol K. at constant temperature, the volume of a
given amount of gas is V . at constant pressure, the volume of a given
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