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A) H_2SO_4 B) SO_4^{2-} C) H_3SO_4^+
D) HSO_4^+ E) OH^- 2) The Page 5/31

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compounds as primary,
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OH a) b) OH OH OH OH c) d) OH
OH OH OH e) f) OH OH OH 2. Fill
in the blanks in the following
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mole of Li atoms = 7 g (b) 1 mole

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of Na atoms = 23 g (c) 1 mole of
Ca atoms = 40 g (d) 1 mole of Fe
atoms = 56 g (e) 1 mole of Ag
atoms = 108 g (f) 1 mole of Pb
atoms = 207 g W9.2 (a) 1 mole of
Cl atoms = 35.5 g 10 moles of Cl
atoms = 35.5×10

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