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Sample Study Sheet 14.1: Electronegativity, Types of Chemical Bonds, and Bond Polarity Memorize the following. Be sure to check with your instructor to determine how much you are expected to know of the following. The information on Table 14.1

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Chapter 14

krcapps. Chemistry Chapter 14 Study Guide. percent by mass. percent by volume. dilution. Molality (m) mass of solute/ mass of solvent x 100. volume of solute/volume of solution x 100. $M_1V_1=M_2V_2$.

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14. 100 cm. 15. Base units are defined units based on specific objects or events in the physical world. Derived units are defined by combining base units. 16. Density is a ratio that compares the mass of an object to its volume. 17. ... Study Guide - Chapter 2 Analyzing Data ...

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Chemistry II: Chapter 14 Mixtures and Solutions. a type of heterogeneous mixture whose particles settle out over time and can be separated from the mixture by filtration. a heterogeneous mixture of intermediate-sized particles (between atomic-size of solution particles and the size of suspension particles).

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