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Three different solutes, sucrose ( $C_{12}H_{22}O_{11}$ ),  $NaNO_3$  and  $MgCl_2$  are used

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to prepare solutions. Each solution is prepared by dissolving 0.01 mol of solute in 1000g of water. Which answer choice lists in order from highest to lowest vapor pressure?  $C_{12}H_{22}O_{11}$ ,  $NaNO_3$ ,  $MgCl_2$

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Chapter 16 Solutions167 SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471-477) This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves. Solution Formation (pages 471-472)

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