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molar mass of  $C_6H_6O = 94.11 \text{ g}$  21. a. molar mass of  $SO_3 = 80.07 \text{ g}$  49.2 mg  $SO_3$  1 g 1000 mg 1 mol 80.07 g = 6.14  $10^{-4}$  mol  $SO_3$  b. molar mass of  $PbO_2 = 239.2 \text{ g}$  7.44 x 10<sup>4</sup> kg  $PbO_2$  1000 g 1 kg 1 mol 239.2 g = 3.11 10<sup>5</sup> mol  $PbO_2$  c. molar mass of  $CHCl_3 = 119.37 \text{ g}$  59.1 g  $CHCl_3$  1 mol 119.37 g = 0.495 mol  $CHCl_3$  d. molar mass of  $C_2H_3Cl_3 = 133.39 \text{ g}$  3.27 ...

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6: An Overview of Organic Reactions. This chapter is designed to provide a gentle introduction to the subject of reaction mechanisms. Two types of reactions are introduced—polar reactions and radical reactions. The chapter briefly reviews a number of topics you should be familiar with, including rates and equilibria, elementary thermodynamics and bond dissociation energies.

*6: An Overview of Organic Reactions - Chemistry LibreTexts*

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