

Chapter 7 Chemistry

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7. Write a balanced chemical equation for each chemical reaction: a) Solid zinc reacts with aqueous hydrochloric acid (HCl) to form zinc chloride and hydrogen gas. b) Gaseous methane (CH₄) reacts with oxygen gas to form carbon dioxide gas and liquid water.

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252Chapter 7 Energy and Chemical Reactions 1 Although chemists recognize that matter can be converted into energy and energy into matter, this matter-energy conversion is small enough to be disregarded. 7.1 Energy253 objeCtive7 objeCtive5 objeCtive7

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NCERT Solutions for Class 12 Chemistry Chapter 7 - The p-Block Elements The p-Block Elements consists of groups 13 to 18 of the periodic table. Chapter 7 chemistry class 12 is a continuation of class 11 syllabus. While the course of grade 11 deals with the first two groups of p-Block, 13 and 14; class 12 syllabus deals with groups 16 to 18.

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