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7 quantum theory 1 41
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Theory and the
Electronic Structure of
Atoms Page 133 20. If a
hydrogen atom and a

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helium atom are traveling at the same speed, A) the wavelength of the hydrogen atom will be about 4 times longer than the wavelength of the helium atom. B) the wavelength of the hydrogen atom will be about 2 times longer than the

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arose in Rutherford's

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nuclear model. A nucleus and electron attract each other; to remain apart the electron must move. • The energy of the electron ' s movement must balance the energy of attraction. • Physics established that a charged particle moving in a curved

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Problems (Chapter 7):

Chrome
Quantum Theory

CHEM 1A 1. Hydrogen

has a red emission line

at 656.3 nm, what is the

energy and frequency of

a photon of this light?

Note: Planck's

constant = 6.626×10

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– $34 \text{ J} \cdot \text{s}$, speed of
light = $2.998 \times 10^8 \text{ m/s}$

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7-8 The Quantum
Theory of Energy Any
object (including atoms)
can emit or absorb only
certain quantities of
energy. Energy is
quantized; it occurs in

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fixed quantities, rather than being continuous. Each fixed quantity of energy is called a quantum. Energy can be gained or lost only in whole number multiples of. An atom changes its energy state by emitting or absorbing one or more quanta of ...

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Problems - 7.1 - 7.7

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Follow-up Problems 7.1

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