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Chemical reactions and  
stoichiometry | Chemistry | Khan

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Academy Stoichiometry Basic

Introduction, Mole to Mole, Grams  
to Grams, Mole Ratio Practice

Problems Importance of

Chemistry in Life, Everyday Uses -

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Introduction The Periodic Table:

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Pass Chemistry How Small Is An  
Atom? Spoiler: Very Small.

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The Origin of the Elements

Balancing Chemical Equations -

Chemistry Tutorial ~~Dalton's~~

~~Atomic Theory | #aumsum #kids~~

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Magic Number Method

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to Grams Hydrocarbon Power!:  
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Limiting Reagent and Percent  
Yield The Electron: Crash Course  
Chemistry #5 Balancing Chemical

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Equations Practice Problems

Introduction to Oxidation

Reduction (Redox) Reactions 9

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Intro to Chemistry, Basic

Concepts - Periodic Table,

Elements, Metric System \u0026amp;

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Chemistry \u0026 Solve Problems

Acids Bases and Salts Chapter 9

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CHAPTER 9 REVIEW.

Stoichiometry. SECTION 1. SHORT

ANSWER Answer the following

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questions in the space provided.

1. \_\_\_\_\_ The coefficients in a chemical equation represent the.
- (a) masses in grams of all reactants and products.
  - (b) relative number of moles of reactants and products.
  - (c) number of atoms of each

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element in each compound in a ...

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Answers

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Stoichiometry CHAPTER 9 REVIEW

Stoichiometry SECTION 3

PROBLEMS Write the answer on

*Page 14/46*

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the line to the left. Show all your work in the space provided. 1.

\_\_\_\_\_ The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of  $N_2$  are mixed with 12.0 mol of H

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SECTION 3 PROBLEMS Write the answer on the line to the left.

Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the



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theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of  $N_2$  are mixed with 12.0 mol of  $H_2$  according to the following equation:  $N_2(g) + 3H_2(g) \dots$

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CHAPTER 9 REVIEW stoichiometry

(which you studied in Chapter 3)

deals with the mass relationships  
of elements in

compounds. Reaction

stoichiometry involves the mass

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## Review Stoichiometry

relationships between reactants and products in a chemical reaction.

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Section 1 Answer Key

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82 SHORT ANSWER Answer the

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Key Concepts: Terms in this set  
(12) Composition Stoichiometry .  
calculations involving the mass  
relationships of elements in  
compounds. Reaction  
Stoichiometry. calculations  
involving the mass relationships  
between reactants and products



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in a chemical reaction. Mole Ratio

...

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### Stoichiometry Sections 1-3

Introduction to Stoichiometry

Ideal Stoichiometric Calculations

Limiting Reactant and Percent

Yield 2 Definitions. Composition

stoichiometry deals with the mass

relationships of elements in

compounds. Reaction

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stoichiometry involves the mass relationships between reactants and products in a ...

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Composition stoichiometry.  
Reaction stoichiometry. Mole  
ratio. Limiting reactant.

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calculations involving the mass relationships of elements in c...  
calculations involving the mass relationships between reactant...  
a conversion factor that relates the amounts in moles of any t...  
the substance that controls the quantity of product that ...

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Reaction stoichiometry uses  
molar relationships to determine  
the amounts of unknown

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## Review Stoichiometry

reactants or products from the amounts of known reactants or products.

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Section 2 Answers

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The following pages contain the bulk (but not all) of the information for the chapter 9 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you...

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