

Chapter Test B Arrangement Of Electrons In Atoms

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Modern Chemistry 31 Chapter Test Chapter: Arrangement of Electrons in Atoms PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. The principal quantum number of an electron is 4. What are the

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Modern Chemistry 28 Chapter Test Name Class Date Chapter Test A, continued _____ 8. What is the frequency of a photon whose energy is 3.4×10^{-19} J? ($h = 6.626 \times 10^{-34}$ J-s) a. 8.8 $\times 10^{26}$ Hz b. 5.1 $\times 10^{14}$ Hz c. 1.9 $\times 10^{-15}$ Hz d. 2.3 $\times 10^{-52}$ Hz _____ 9. When electromagnetic radiation strikes the surface of a metal, electrons are ejected from the metal's ...

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(b) particle nature. (d) wave nature. 6. c According to Bohr, which is the point in the figure below where electrons cannot reside? (a) point A (c) point C (b) point B (d) point D 7. c According to the quantum theory, point D in the above figure represents (a) the fixed position of an electron. (b) the farthest position from the nucleus that an ...

4 Arrangement of Electrons in Atoms

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_____ refers to the arrangement of conditions in a study so that meaningful comparisons of the effect of the IV can be made 1. intervention 2. experimental design 3. single-subject design 4. group comparison design

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