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energy of He would be about  $4 \times$  (twice the charge and half the distance) the ionization energy of H if this were the case.  
2. Several possible answers. A typical one would have all three electrons at a much farther distance than in H from the

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ChemActivity 55 & 56 Gibbs Energy and Equilibrium •

Know that the Gibbs energy was developed as a single energy term that incorporates both the enthalpy and entropy effects on reaction spontaneity.  $G^\circ = H^\circ -$

$T S^\circ$  • Be able to calculate  $G^\circ_{\text{rxn}}$  from the above equation (make sure your units match! kJ J)

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