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-Answer Key. Balance the equations

below: 1) $1\text{N} + 3\text{H}_2 \rightarrow 2\text{NH}_3$ 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$

3. Balancing Chemical

Equations Answer Key Tips for

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Balancing Equations When balancing equations, remember chemical reactions must satisfy conservation of mass. Check your work to make Page 1/3

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Balancing Chemical Equations

–Answer Key. Balance the equations below: 1) $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$. 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$

3. Balancing Chemical

Equations Answer Key Tips for

Balancing Equations When balancing equations, remember chemical reactions must satisfy conservation of mass. Check your work to make certain you have the same number and type of atoms on the

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