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Balancing Chemical Equations Part 2

Chemical Equations Reactions Section
2

1. They could draw a picture of the
chemical reaction. 2. They could write
a word equation for the chemical

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Reaction: "Two molecules of hydrogen gas react with one molecule of oxygen gas to produce two molecules of water vapor." 3. They could write the equation in chemical shorthand.
$$2 \text{H}_2 (\text{g}) + \text{O}_2 (\text{g}) \rightarrow 2 \text{H}_2\text{O} (\text{g})$$

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8.2: Chemical Equations - Chemistry
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Chemical Equations and Reactions
SECTION 2 SHORT ANSWER

Answer the following questions in the
space provided. 1. Match the equation

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type on the left to its representation on the right.

c synthesis (a) $AX + BY \rightarrow AY + BX$

d decomposition (b) $A + BX \rightarrow AX + B$

b single-displacement (c) $A + B \rightarrow AX + A$

a double-displacement (d) $AX \rightarrow A + X$

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Here is the balanced symbol equation:
 $2\text{Cu} + \text{O}_2 \rightarrow 2\text{CuO}$. You can see that
we now have two copper atoms and
two oxygen atoms on each side. This
matches what happens in the reaction:

Chemical equations - Types of

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Reaction - KS3 Chemistry ...

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Reactions SECTION 2 Name Class

Date Chemical Formulas and

Equations continued HOW TO

BALANCE A CHEMICAL EQUATION

To balance an equation, you must use coefficients. A coefficient is a number

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that is placed in front of a chemical formula. For example, 2CO represents two carbon monoxide molecules.

2 SECTION 2 Chemical Formulas and
Equations

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Equations continued CHECKING

SYMBOLS When you write a chemical formula, it is important that you check to make sure that it is correct. If you use the wrong formula or symbol in an equation, the equation will not

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describe the correct reaction. Even a
small mis-

14 SECTION 2 Chemical Formulas
and Equations

This is second lecture on a CBSE
Class 10th topic - Chemical Reactions

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and Equations. This chapter is divided into 3 lectures, so make sure you watch the oth...

Chemistry: Chemical Reactions and
Equations (Part 2 ...
the general equation for a single

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displacement reaction is displacement of the metal in a compound by another metal, displacement of the halogen in a compound by another halogen, displacement of the hydrogen in water by a metal species, the displacement of a hydrogen in acid by a metal species

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chapter 8 chemical equations and
reactions section 2 ...

Section 2: Chemical equations.

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in this set (13) chemical equation. are

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shorter, easier, way to show chemical reactions, using symbols instead of work. What's needed to write an equation - chemical formulas, reactants, and products. chemical ...

Section 2: Chemical equations

Page 24/41

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4.1: Chemical Reactions and Chemical
Equations Last updated; Save as PDF

Page ID 24181; No headers $[2H_2 + O_2 \rightarrow 2H_2O]$

Chemical formulas and other symbols are used to indicate the starting materials, or reactants, which by convention are

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Written on the left side of the equation,
and the final compounds, or products,
which are written on the right.

4.1: Chemical Reactions and Chemical
Equations - Chemistry ...

Types of reaction Atoms are

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Rearranged during chemical reactions, and are not lost or gained. Chemical reactions can be represented using equations. Catalysts speed up reactions without being used...

Chemical reactions - Types of reaction

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KS3 Chemistry ... Section 2

Ch. 8 Chemical Equations And

Reactions 1. Chemical Equations and

Reactions CHAPTER 8 2. What is a

chemical equation? 3. List and

describe some indications of an

occurrence of a chemical reaction. 4.

List and describe some indications of

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an occurrence of a chemical reaction.

Answers

Ch. 8 Chemical Equations And
Reactions

CHEMICAL EQUATIONS AND
REACTIONS 261 SECTION 1 O

OBJECTIVES List three observations

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that suggest that a chemical reaction has taken place. List three requirements for a correctly written chemical equation. Write a word equation and a formula equation for a given chemical reaction. Balance a formula equation by inspection.

Describing Chemical ...

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CHAPTER 8 Chemical Equations and
Reactions

Chemical Equations and Reactions

SECTION 2 SHORT ANSWER

Answer the following questions in the
space provided. 1. Match the equation

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type on the left to its representation on the right. a b synthesis decomposition single-displacement double-displacement (a) $AX + BY \rightarrow AY + BX$ (b) $A + BX \rightarrow AX + B$

Where To Download Chemical Equations

Chemical reactions are represented by balanced chemical equations, with chemical formulas symbolizing reactants and products while chemical equations give information in two major areas: first, they tell us what substances are reacting (those being used up) and what substances are

Where To Download Chemical Equations Reactions (those being made). Answers

Chemical Reactions and Chemical
Equations - Owlcation ...
Section 2 Chemical Formulas and
Equations Key Concept Chemical
formulas and chemical equations are

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Used to show how atoms are rearranged to form new substances in a chemical reaction. □ Chemical formulas are a simple way to describe which elements are in a chemical substance. □ Chemical equations are a concise way to write how atoms are

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Section 2 Chemical Formulas and Equations

The five general types of reaction are combination, decomposition, single-replacement, double-replacement, and combustion. Not all chemical reactions fit uniquely into only one category.

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Occasionally, a reaction may fit equally well into two categories. Nevertheless, recognizing a reaction as a particular type is useful.

11.2 Types of Chemical Reactions -
Henry County School ...

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It can be balanced by putting a 2 before O₂, giving the balanced equation: CH₄ + 2 O₂ → CO₂ + 2 H₂O. This equation does not have any coefficients in front of CH₄ and CO₂, since a coefficient of 1 is dropped.

Matrix method. Generally, any chemical equation involving J different

Where To Download Chemical Equations Molecules can be written as: Answers

Chemical equation - Wikipedia
- Finish Predicting Reactions- w
orksheet #1 (* ANSWERS) *Extra
Practice: Combustion Read 3.3 (p. 137
- 141), ANSWER p. 141 #41 - 50:

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Wed, Oct 9: Continue work from
yesterday: Predicting Reactions *
Worksheet #2 - Predicting the
Products of Chemical Reactions
SUMMARY- Types of Decomposition
Reactions (*HANDOUT)

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