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A compound is found to have a molecular mass of 90 atomic mass units and simplest formula of  $C_2H_5O$ . The molecular formula of the substance is: \*\*Use atomic masses of  $C = 12$  amu,  $H = 1$  amu,  $O = 16$  amu\*\* A.  $C_3H_6O_3$  B.  $C_4H_{26}O$  C. C

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Answers  
4 H 10 O 2 D. C 5 H 14 O

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DIF: L1 REF: p. 253 OBJ: 9.1.1 Identify

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the charges of monatomic ions by using the periodic table, and name the ions.

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Answers and other study tools. ... A reaction where a compound breaks down into elements and or smaller compounds. ... animal farm studyguide answers chapters 1-4. 27 terms. jkhampt4u. Unit 4: Chemical Bonding Test Review. 38 terms.

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## Answers - Quizlet

Answers. 1. magnesium sulfate. 2. lead (II) iodide. 3. iron (III) oxide. 4. chromium (III) hydroxide. 5. ammonium chloride.

## Naming Ionic Compounds Test

## Questions - ThoughtCo

## CHEMICAL FORMULA MULTIPLE

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**CHOICE TEST.** 1. It is used to express the composition of molecules and ionic compounds in terms of chemical symbol.

2. What do you call the formulas that give the true number of atoms.

3. The charge of chemical formula should be. a. positive. b. negative.



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## **CHEMICAL FORMULA MULTIPLE CHOICE TEST - Blogger**

Quiz Flashcard. Revision of Chemical Formulas This 10-item test will measure how well you understood the rules in writing down the correct chemical formulas of compounds. You have 5 minutes to complete the test. Identify the

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Answers  
correct chemical formula of the given compound. Tick the letter that corresponds to your answer.

## **Year 10 Chemical Formulas Test - ProProfs Quiz**

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Answers simplest formula of  $C_2H_5O$ .

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The molecular formula of the substance is:

\*\*Use atomic masses of C = 12 amu, H = 1 amu, O = 16 amu\*\* A. C<sub>3</sub>H<sub>6</sub>O<sub>3</sub> B. C<sub>4</sub>H<sub>26</sub>O C. C<sub>4</sub>H<sub>10</sub>O<sub>2</sub> D. C<sub>5</sub>H<sub>14</sub>O

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Questions Start studying Chapter 7 -

Chemical Formulas and Chemical Page

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## **Chemical Formulas Compounds Test Answers**

Chemical Compounds Practice Quiz. This online quiz is intended to give you extra practice in naming compounds, writing formulas and calculating molar masses (formula weights). Select your preferences

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Answers below and click 'Start' to give it a try!

Number of problems: 5 10 25 50 100!

Quiz type: Names & formulas

## **Chemical Compounds Practice Quiz | Mr. Carman's Blog**

List of Chemical Compounds, Their  
Common Names and Formulas Chemical

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**Compound.** A chemical compound is a chemical substance composed of many identical molecules composed of atoms from more than one element held together by chemical bonds. There will be a fixed ratio for a chemical compound determining the composition of it.

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## List of Chemical Compounds and their Common Names and Formulas

16. The chemical formula for iron(II) nitrate is A)  $\text{Fe}_2(\text{NO}_3)_3$  B)  $\text{Ir}(\text{NO}_2)_2$  C)  $\text{Fe}_2\text{N}_3$  D)  $\text{Fe}(\text{NO}_3)_2$  E)  $\text{Fe}(\text{NO}_2)_2$
17. Name the compound  $\text{Co}_2(\text{SO}_3)_3$ .  
A) cobalt sulfite B) cobalt(II) sulfite C) cobalt(II) sulfate D) cobalt(III) sulfite E)

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cobalt(III) sulfate

**Nomenclature Practice Quiz. Answers are on the last page ...**

Chemical Formulas and Chemical Compounds SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1. c In a Stock



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Answers system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

## **7 Chemical Formulas and Chemical**

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## Answers

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## Chemistry Chapter 7 Chemical Formulas and Compounds ...

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The polyatomic anion perxenate, which has a chemical formula of  $\text{XeO}_6^{4-}$  has a charge of -4, meaning that it has an excess of four valence electrons aside from the valence ...

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