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- MAOP Solid

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calcium carbonate is commonly used in antacids because it reacts with the hydrochloric acid found in the stomach. The products of this reaction are aqueous calcium chloride, carbon dioxide, and water.

(s) + + 142-0 (Q)

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commonly used in

antacids because it

reacts with the

hydrochloric ound

in the stomach. The

products of this

reaction are

aqueous calcium

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chl.cle, carbon
dioxide, and water.
(s) + + 142-0 (Q)

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Ethane C₁₄H₁₀Cl₄.

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Decomposition

reaction Is a type of
chemical reaction

where one reactant
yields two or more

products. It is just

an opposite of a

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synthesis or
combination
reaction.

Displacement

reaction is a type of
reaction in which a
less reactive
element is displaced
by a more reactive
element in a

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dioxide, and water.

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