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18 SOLUTIONS MANUAL Section 18.3

Hydrogen Ions and pH pages 650 – 658

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The concentration of either the H<sup>1</sup> ion or  
the OH<sup>2</sup> ion is given for four aqueous

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solutions at 298 K. For each solution, calculate  $[H^+]$  or  $[OH^-]$ . State whether the solution is acidic,

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~~MULTIPLE CHOICE QUESTIONS~~

~~NOTE: A table of standard reduction potentials is required to work many of these problems.~~  
1. Complete and balance the following redox equation. When properly balanced using the smallest whole-number coefficients, the coefficient of S is  
 $\text{H}_2\text{S} + \text{HNO}_3 \rightarrow \text{S} + \text{NO}$  (acidic solution)  
2.

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