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A mixture is made up of two or more substances

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chemistry. Describe the difference between mass and weight. Describe the difference between chemical and physical change. Describe the difference between accuracy and precision. Describe the five states of matter, and give an example of each

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AP Chemistry Practice Test #1 - Key Chapters 1, 2,

and 3 1. In 1928, 3.3 g of a new element was isolated from 660 kg of the ore molybdenite. The percent by mass of this element in the ore was: a. 5 % b. 6.6 % c. 3.3 % d.  $5.0 \times 10^{-4}$  % e. 2.2 %

2. A quantitative observation a. contains a number and a unit. b. does not contain a number.

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1 MCQ Test With Answer for Chemistry chapter 1

(Fundamental Concepts) In this topic, Student should be

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able to : a) Define relative atomic, molecular and formula masses, based on the 12C scale and concept of isotopes. b) Explain mole in terms of the Avogadro ' s constant.

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