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set (53) the idea of arranging the elements in a table according to their chemical and physical properties is attributed to _____.

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conductors of electric current and heat.

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Section 6.1 Development of the Modern Periodic Table. 1.octaves.

2.eight. 3.nine. 4.accepted. 5. Dmitri Mendeleev. 6.atomic mass.

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arranged elements according to
atomic mass and used the
arrangement to predict the properties
of missing elements. elements that are
characterize by the filling of p orbitals
are classified as _____. which subatomic
particle plays the greatest part in
determining the properties of an
element.

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reactions for which electrons are
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reactions).

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Describe the five states of matter, and
give an example of each

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