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determined the value of x in the formula of hydrated ethanedioic acid,

$\text{HOOC}-\text{COOH}\cdot x\text{H}_2\text{O}$,
by titrating a known mass of the acid with a

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0.100 mol dm⁻³ solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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electrons :

aluminium ions 3

: 1 amount Al 13

1 1.296 10 C

96500Cmol 3 «»

4.48 107 «mol»

mass Al « 4.48

107 mol 26.98 g

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«g» Award [3]
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