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A change in physical form of a substance or physical properties of a substance like color, melting point, state, size, shape, and arrangement of molecules is known as physical change. A change in chemical properties of a substance like composition, reactivity with water,

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corrosiveness, and pH is known as
chemical change. In chemical change,
a new substance will be formed.

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35. Aqueous solutions of potassium iodide and silver nitrate are mixed, forming the precipitate silver iodide.
chemical equation: $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$
complete ionic equation: $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$

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Chemical Reactions

Solution. Physical: boiling and melting are physical changes. When water boils, no bonds are broken or formed. The change could be written:



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$\text{H}_2\text{O}(\text{g})$ Chapter:

The dark grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

3.6: Changes in Matter: Physical and Chemical Changes ...

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30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation,

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particles in a liquid gain enough

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States of Matter

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