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Mole. Commonly

abbreviated mol,

SI base unit

used to measure

the amount of a

substance.

Avogadro's

Number. $6.02 \times$

10^{23} Number of

representative

particles in a

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mole of

anything. Molar

mass. Mass in

grams of one

moles of any

pure substance.

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involve changing

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substances

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(reactants
change to one or
more products).

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chemical

equation to

convey as much

information as

possible about

what happens in

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a chemical
reaction.

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of the reactants

and the mass of

the products in

a chemical

reaction, and

explain how

these masses are

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related. The
coefficients in
the balanced
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indicate the
molar
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between each
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used in many
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alloys because
it is extremely
strong and
lightweight.

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tetrachloride (
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titanium oxide (TiO_2) using chlorine and coke (carbon).

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amounts of
reactants used
and the amounts
of products
formed in a
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