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4NH (g) 5O (g) 4NO(g) 6H O(g) kJ mol 2 2+ + =? ?HÖ 910 ?1

Describe the effect, if any, of each of the following changes

on the equilibrium concentration of nitrogen monoxide in a

particular equilibrium mixture, giving a reason in each case.

(i) Increasing the pressure, at constant temperature [2]

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– 2 – Section A Answer all questions. Write your answers in

the boxes provided. 1. Ethanedioic acid is a diprotic acid. A

student determined the value of x in the formula of hydrated

ethanedioic acid, HOOC?COOH•xH<sub>2</sub>O, by titrating a known

mass of the acid with a 0.100 mol dm<sup>-3</sup> solution of NaOH

(aq). 0.795 g of ethanedioic acid was dissolved in distilled

water and made up to a total ...

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Ethane, C<sub>2</sub>H<sub>6</sub>, and disilane, Si<sub>2</sub>H<sub>6</sub>, are both hydrides of group 4 elements with similar structures but with different chemical properties. (a) (i) Deduce the Lewis (electron dot) structure for Si<sub>2</sub>H<sub>6</sub> showing all valence electrons. [1] (ii) State and explain the H–Si–H bond angle in Si<sub>2</sub>H<sub>6</sub>. [2]

## CHEMISTRY Candidate session number STANDARD LEVEL PAPER 2

[2] (b) Marit arranged the values she found in Table 12 into an energy cycle. C<sub>2</sub>H<sub>4</sub>(g) + H<sub>2</sub>(g) C<sub>2</sub>H<sub>6</sub>(g) 3 O 2 2CO<sub>2</sub>(g) + 3H<sub>2</sub>O – 1 4 1 1 k J m o l – 1 – 2 8 6 k J m o l – 1 – 1 5 6 0 k J m o l – 1 (hydrogenation) (l) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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