

## Chemistry Stoichiometry Packet Answers

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~~Stoichiometry 6: Limiting Reactant \u0026 Excess Reactant~~

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chemistry is number of units in a mole. The number of

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Chapter 12 Stoichiometry Packet Answer Key Answer:  $4.93 \times 10^{-5}$  L or 49.3  $\mu$ L In Example 12.2.1 and Example 12.2.2, the identity of the limiting reactant has been apparent: [Au(CN) 2 ] ? , LaCl 3 , ethanol, and para -nitrophenol.

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## *Stoichiometry Packet Answers Chapter 12*

Chemical Reactions and Stoichiometry. Given the equation  $3A + B \rightarrow C + D$ , you react 1 mole of A with 3 moles of B. True or false: B is the limiting reactant because you have fewer moles of B than A.

Answer: False.  $\text{NaHCO}_3$  is the active ingredient in baking soda. How many grams of oxygen are in 0.35 g of  $\text{NaHCO}_3$ ? a) 0.012 g. b) 0.022 g. c) 0.067 g. d) 0.20 g

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