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An Introduction to Thermal Physics

1. DIESEL ENGINES 2. As in the Otto cycle, the red and blue curves are adiabats, so no heat is exchanged during these steps. The efficiency of the Diesel cycle is most easily

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worked out by calculating the heat input Q_1 along the yellow step 2 to 3 and the heat expelled during the green step 4 to 1. The efficiency is then $e = W/Q_1$.

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in Schroeder), $\Delta H_{H_2O} = 285.83 \text{ kJ mol}^{-1}$ and $\Delta H_{H_2} = 890.36 \text{ kJ mol}^{-1}$ (12) This time, the final volume is $1/3$ of the initial volume, since the 2 moles of water has condensed out as liquid with negligible volume compared to the gases. Thus the atmosphere does work $PdV = RT \ln V_2/V_1$ (13) = $8.31 \text{ J K}^{-1} (298 \text{ K}) (\ln 2)$ (14) =

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