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27 Questions Show answers. Question 1 . SURVEY . 30 seconds . Q. What atom matches this electron configuration? $1s^2 2s^2 2p^6 3s^2$. answer choices . Neon. Magnesium. Aluminum. ... The electron configuration of an atom is $1s^2 2s^2 2p^6$. The number of valence electrons in the atom is . answer choices . 3. 6. 8. 10. Tags: Question 14 . SURVEY ...

Electron Configurations | Periodic Table Quiz - Quizizz
Preview this quiz on Quizizz. What atom matches this electron configuration? $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^{10}$... 33

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Q. How many electrons can an p sublevel hold? ... What electron configuration matches an oxygen atom? answer choices . $1s^2 2s^2 2p^6 3s^2, 3p^6 4s^2 3d^{10} 4p^5$. $1s^2 2s^2 2p^4$.

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How many valence electrons does it have? 2. Do a sketch of its atom (Bohr model) showing the electron shells) Quantum Mechanical Model of the Atom-The electron does NOT have an exact orbit around the nucleus (like bohr ' s model) Atomic orbital-Electron Configuration- composed ...

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Solution for Consider the electron configuration $1s^2 2s^2 2p^6$

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3s²3p⁶3d⁸4s². Which of the following statements is TRUE?
(A) There are 14 electrons on the third...

Answered: Consider the electron configuration... | bartleby
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Electron configurations for the third and fourth periods.
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Valence electrons. Valence electrons and ionic compounds.

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To answer the question, you need to know the electron configuration of the atoms to see which one has 7 unbonded electrons, like chlorine. Fluorine, element number 9, has 2 electrons in the s sublevel (K shell). The L shell is incompletely filled, with 7 electrons. 9.

Atomic Structure Chemistry Quiz - ThoughtCo

38. The ground state electron configuration of Fe is _____.

A. $2 1s 2s 2 3s 3p^6 3d^6$ B. $2 1s 2 2s 2p^6 3s^2 3p^6 4s 3d^6$ C. $2 1s^2 2s 2p^6 3s^2 3p^6 4s^2$ D. $2 1s 2 2s 2p^6 3s 3p^6 4s^2 4d^6$ 39.

The ground state electron configuration of Ga is _____. A. $2 1s 2s^2 3s^2 3p^6 3d^{10} 4s^2 4p^1$ B. $6 1s 2 2s 2p 3s^2 3p^6 4s^2 4d^{10}$

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4p1 C. 2 1s2 2s 2p6 3s2 3p6 4s2 3d10 4p1

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practice problems and step-by-step solutions. Write out the orbital box diagram and the...

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Which of the following is the correct electron configuration for the bromide ion, Br⁻? a) [Ar] 4s² 4p⁵. b) [Ar] 4s² 3d¹⁰ 4p⁵. c) [Ar] 4s² 3d¹⁰ 4p⁶. d) [Ar] 4s² 3d¹⁰ 4p⁶ 5s¹. e) [Ar] 4s² 3d¹⁰ 3p⁶. answer . text explanation. 6-89 . Which of the following elements has the largest number of electrons for which the principal quantum ...

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