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A covalent bond in which two atoms share three pairs of electrons is called a(n) bond. 6. A bond in which two atoms share one pair of electrons is a bond. 7. A nonbonding pair of electrons in the valence shell of an atom is called a(n) pair. Answer the following items in the space provided. 8.

[Complete each statement below by choosing a term from the ...](#)

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structures 2. b A covalent bond consists of (a) a shared electron. Section 6 1 Introduction To Chemical Bonding Answer Key CHAPTER 6 Chemical Bonding Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1.

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Concept Review: Covalent Bonds 1. A covalent bond forms when two or more valence electrons are attracted by the positively charged nuclei of two atoms and thus are shared between both atoms. 2. The H<sub>2</sub> molecule is stable because each hydrogen atom now has a shared pair of electrons and has achieved a stable noble gas configuration. 3.

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Holt Chemistry Concept Review Answer Key Chapter12 Summary. 2 × atomic mass of carbon = 2atoms(12.011amu atoms) = 24.022amu. 6 × atomic mass of hydrogen = 2atoms(1.0079amu atoms) = 6.0474amu. 1 × atomic mass of oxygen = 1atoms(15.9994amu atoms) = 15.994amu.

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each statement or best answers each question. 1. Chemical bonding in metals is a. the same as ionic bonding. b. the same as covalent bonding. c. a combination of ionic and covalent bonding. different from ionic or covalent bonding. 2. he valence electrons in a metallic bond move freely throughout the network of metal atoms.

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A covalent bond forms when two or more valence electrons are attracted Holt Chemistry Concept Review Holt Chemistry ... Holt Chemistry Concept Review - trumpetmaster.com a concept of chemical bonding theory that is based on the assumption that atoms tend to have either empty valence shells or full valence shells of eight electrons.

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