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marks) Complete the following
table. Pathogen

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answer: $q = m \times C \times \Delta T$ $q =$
 $210 \times 4.18 \times 65$. $q = 57057$. q
 $= 57.057$ kJ . Now . H. Comb =
 q / n . $57.057 / n$. C = 2. H = 5. OH =
1367. $n = 0.04174$ mol. C = 2. H = 5.
OH . m . C = $n \times MM$. 2. H = 5. OH
C = 2. H = 5. OH . m . C = 0.04174

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$\times 46.068. 2. H. 5. OH \quad m. C. 2.$
 $H. 5. OH = 1.923 \text{ g} .$ Now 50% of
heat has been lost to surroundings.
initial . $m = 2 \times 1.920 \text{ g} . C. 2.$
H. 5

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