

## Introduction To Solubility Phet Lab Answers Key

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~~General Chemistry Lab: Solubility Product Constant of Silver Acetate~~

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Description. Add different salts to water, then watch them dissolve and achieve a dynamic equilibrium with solid precipitate. Compare the number of ions in solution for highly soluble NaCl to other slightly soluble salts. Relate the charges on ions to the number of ions in the formula of a salt. Calculate Ksp values.

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What happens when sugar and salt are added to water? Pour in sugar, shake in salt, and evaporate water to see the effects on concentration and conductivity. Zoom in to see how different sugar and salt compounds dissolve. Zoom in again to explore the role of water.

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~~pHet Simulations~~ → "Concentration", "Salts and ...

Visualizing ionic formulas using Salts and Solubility simulation from the PhET Activity 1 Learning Goals Students will be able to: • Determine the chemical formula by observation of ionic ratios in solutions • Relate the simulation scale to real lab equipment through illustration and calculations

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