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states that all
matter is made
of small

particles that
are in random
motion and that
have

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average kinetic

energy of an-
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object's atoms
or molecules B.

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-the sum of the
kinetic and
potential energy
of all the atoms
in an object -
as temperature
increases.

A..hermal energy
- S 2.

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1. kinetic theory Of matter
2. plasma
3. crystals
4. solid
5. amorphous material
6. steam
7. _e_

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thermal expansion 8.

liquid 10. ice

use the words in

the box to fill

in the blanks.

Column I water

vapor b. state

of matter with

no definite

shape but with

definite volume

c. solid which

is not made of

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crystals d.

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to liquefy at
the melting
point, as the
particles gain
enough energy to
overcome their
ordered
arrangement.

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Theory. Explains
how the
particles in
gases behave.

Plasma. Gas like
mixture of
charged

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particles.

Crystals.

Particles

arranged in

repeating ...

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