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$1.2C(s) + O_2(g) \rightarrow 2CO(g)$ 1 mol O_2 2 mol C 2 mol CO / 2 mol C 2 mol $C/1$ mol O_2 2 mol $CO/1$ mol O_2 2 mol $C/2$ mol CO 1 mol O_2 / 2 mol CO 2 2. $WO_3(s) + 3H_2(g) \rightarrow W(s) + 3H_2O(l)$

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