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Paper 02 – Structured Essays

Question 1 Syllabus References: A:

7.1, 7.7, 7.3, 7.4; B2 7.1, 7.2, 7.3 Part

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(a) (i) tested candidates' knowledge of experimental procedure for collecting and measuring the volume of gas generated during a chemical reaction. Candidates were required to draw a suitable

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This question was attempted by majority of the candidates and the

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performance was fair. In (a)(i)(ii) and (iii) most candidates correctly define standard electrode potential, state factors that affect standard electrode potential and gave uses of the values of standard potential.

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is solution containing 6.22 g of an acid  $H_2Y$  per  $dm^3$ . B contains 3.90 g of NaOH per  $dm^3$  of solution. (a) Put A into the burette and titrate it against 20.0  $cm^3$  or 25.0  $cm^3$  portions of B using methyl orange as indicator. Repeat the titration to obtain consistent titres. Tabulate your

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