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observed evidence that when
M-NM compounds are
dissolved, the metal
particles tend to form
positively charged ions
(cations), while non-metal
particles tend to form
negatively charged ions
(anions).

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Unit 6 - Worksheet 3 Ionic
Compounds Properties Ionic
compounds generally have
higher mp's and bp's than do
molecular compounds. When
molten, they conduct

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electricity. If soluble, the solutions also conduct electricity.

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f) Determine whether substance A and B will sink or float when placed in a bucket of water. A: sink float B: sink float (circle correct response) Defend your answer using the m-V graph, and your outstanding understanding of density. Refer to the table of

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v1.0 3. Next to the formula,
give the name of the
following simple binary
ionic compounds. a. Na_2O b.
 MgCl_2 c. CaBr_2 d. Al_2S_3 e.
 CsBr f. AgF g. Na_3N h. K_2O
5. Write the name of these
ionic substances next to the
formula, using a Roman
numeral to specify the
charge of the cation. ...
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