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barometer . 1. 720 mmHg = 0.95 atm. 2. 760 mmHg = 1 atm. Problems 3 and 4. Calculate the pressure of the gas in the flask connected to the manometer.
3. a. $127 - 84 = 43 \text{ mmHg}$ $730 + 43 = 773 \text{ mmHg}$

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building (using the model kit included in the Lab Kit) a variety of simple molecules and making observations on their structure. Your task will be to develop a better understanding of molecules and their 3-D properties. Procedure: 1) Get out the molecular modeling set from your Supplemental “ Che 106/107

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