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42b. 0.002 037 g e. 1.3 10<sup>2</sup> cm 33c. 350. J f. 1.30 10<sup>2</sup> cm 2. Write the value of the following operations using scientific notation. 10 1 a. 10 3 1 0 1 2 0 6 4 10 2 b. 8 2 3 1 1 0 05 4.3 410 c. 3 10<sup>3</sup> 4.0 10<sup>4</sup> 3. The following data are given for two variables, A and B: AB 18 2 94 66 312 a. In the graph provided, plot the data. inversely ...

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### *2 Measurements and Calculations*

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

### *5 The Periodic Law*

Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. \_\_\_\_ 7. A solution containing 35 g of  $\text{Li}_2\text{SO}_4$  dissolved in 100 g of water is heated from  $10^\circ\text{C}$  to  $90^\circ\text{C}$ . According to information in the figure, this

### *Assessment Chapter Test A - Ed W. Clark High School*

Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20.

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