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on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting

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continued Use this figure to answer
questions 7 and 8. _____ 7. A solution

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containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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continued _____ 13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity.

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_____14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

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13 ...

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Name Class Date Chapter Test B,
continued 18. When a strong acid is
titrated with a weak base, the pH of

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the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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slightly less massive and more stable
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