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proportions, counting atoms  
related terms. Terms in this  
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of mass. states that mass is

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3 neither created nor destroyed during ordinary reactions or physical changes.

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3 definite proportions. law of multiple proportions. atom. States that matter can be neither destroyed nor created during... The fact that a chemical compounds contains the same elements... if two or more different compounds are composed of the same ...

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### Chapter 3 - Chemistry



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Atoms: The Building Blocks  
of Matter. law of  
conservation of mass- mass  
is neither created nor  
destroyed during ordinary  
chemical reactions or  
physical changes. e.g. 20 g  
A + 20 g B 40 g AB.

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CHAPTER 3 REVIEW Atoms: The  
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SECTION 2 SHORT ANSWER

Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the

## 3 Atoms: The Building Blocks of Matter

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of Matter law of  
conservation of mass mass is  
neither created nor  
destroyed during ordinary  
chemical reactions or  
physical changes e.g.  $20\text{ g A} + 20\text{ g B} \rightarrow 40\text{ g AB}$  law of  
definite proportions- a  
chemical compound contains  
the same elements in exactly  
the same proportions by mass  
regardless of the size of

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3 the sample or the source of the compound e.g. If 10 grams of A combine with 20 grams of B to form compound AB, how ...

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...

CHAPTER 3 TEST Class Atoms:  
The Building Blocks of  
Matter MULTIPLE CHOICE On the  
line at the left of each  
statement, write the letter  
of the choice best completes  
the statement or answers the  
question. The behavior of  
cathode rays in a glass tube  
containing gas at low  
pressure led scientists to  
conclude that the rays were  
composed of a. energy

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