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~~Chapter 6 - The~~

~~Electronic Structure of~~

~~Atoms: Part 1 of 10~~ **Fall**

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**Chapter 6 - The**

**Language of**

**Chemistry The**

**Periodic Table:**

**Atomic Radius,**

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**Chemistry**  
**Ionization Energy, and**

**Electronegativity 6.1**

~~Power Point Notes~~

~~Chapter 4:5 Notes~~

*Introduction to Ionic*

*Bonding and Covalent*

*Bonding Chapter 5 2*

Part 1 ~~Pearson Chapter~~

~~6: Section 1: Organizing~~

~~the Elements~~ How to

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10 Scientifically Proven

Study Techniques |

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## Bonding Covalent

## Bonds and Ionic Bonds

## **Lewis Diagrams Made**

## **Easy: How to Draw**

## **Lewis Dot Structures**

CHM 101: Introductory  
Chemistry (Chapter 6)

## 6.2 Covalent Bonding and Molecular

## Compounds Chapter 6—

~~The Electronic Structure  
of Atoms: Part 3 of 10~~

~~Chapter 2—Atoms,~~

~~Molecules, and Ions:~~

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~~Part 1 of 3~~ *Periodic*

*Table Explained:*

*Introduction Ionic*

*Bonding Introduction*

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*Orbitals: Crash Course*

*Chemistry #25* ~~Chapter 6~~

~~The Electronic~~

~~Structure of Atoms: Part~~

~~8 of 10~~ *Zumdahl*

*Chemistry 7th ed.*

*Chapter 6 (Part 2)*

*General Chemistry 1*

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**Chemistry, F.Sc part  
1, Ch#6, Introduction  
to chemical bonding**

*Chemical bond / octet  
rule / Energetics of  
bond formation /*

*Chemistry Part 1 /*

*Chapter 6 Lec 01*

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Discussion Question 5 -

The Rate... ~~Ch 6 Lec 1~~



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~~What is Chemical  
Bond? Chemical  
Bonding FSc Part 1  
Chemistry Periodic  
Trends:~~

~~Electronegativity,  
Ionization Energy,  
Atomic Radius~~

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Electronic Structure of  
Atoms: Part 7 of 10  
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STUDY. PLAY.

Review  
Chemical Bond. A

Chemical Bonding  
mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together.

(Valence cloud) Ionic

Bond. Chemical

bonding that results

from the electrical

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attraction between large numbers of cations and anions. (When metal gives electron ...

Chemical

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STUDY. PLAY.

Electronegativity greater than 1.7 and is a transfer of electrons. The Lower

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Electronegativity

transfers to the higher  
electronegativity) t.

Ionic Bond. Ions pack in  
a crystal lattice

structure. Lattice

Energy. Cation {+} and  
anion {-} create \_\_\_\_\_

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STUDY.PLAY.

Chemical bond. a mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together. ionic bond. chemical bonding that results from the electrical attraction between cations and anions.

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Review CHAPTER 6:

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BONDING The

following pages contain the bulk (but not all) of the information for the chapter 6 test. Focus on this content, but make sure to review class notes, activities,

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handouts, questions, etc.

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document and

NOTHING else, you  
should at least be able to  
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Chemical Bonding

CHAPTER 6 REVIEW

Chemical Bonding

SECTION 3 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

\_\_\_\_\_ The notation for sodium chloride, NaCl, stands for one (a)

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formula unit. (c) crystal.

(b) molecule. (d) atom.

2. \_\_\_\_\_ In a crystal of an ionic compound, each cation is surrounded by a ...

## ~~CHAPTER 6 REVIEW~~

### ~~Chemical Bonding~~

SECTION 6.1 Nature favors arrangements in which potential energy is minimized. For example, a boulder is

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less likely to balance at the top of a hill than it is to roll to the bottom of a valley. The boulder at the top of the hill is not stable. Atoms are usually not stable when isolated. They usually combine to form more stable arrangements of matter.

~~CHAPTER 6~~ Chemical

~~Bonding~~

*Page 20/34*

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## CHAPTER 6 REVIEW

Chemical Bonding

### SECTION 2 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

Use the concept of

potential energy to

describe how a covalent

bond forms between two

atoms. As the atoms

involved in the

formation of a covalent

bond approach each

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other, the

Chapter 6

~~6 Chemical Bonding~~

Review  
The Chemical Bonding

chapter of this Holt

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Chemistry Companion

Course helps students

learn the essential

lessons associated with

chemical bonding.

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~~Chemistry Chapter 6:~~

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~~Chemical Bonding ...~~

1 s<sup>2</sup>2s 2p<sup>6</sup>3 23p<sup>6</sup>4s<sup>1</sup> c.

Gallium 1s 2s 2p<sup>6</sup>3s

3p<sup>6</sup>3d<sup>10</sup>4s 4p<sup>1</sup> d.

Copper 1s 2s 2p

6s<sup>2</sup>3p 3d<sup>10</sup>4s<sup>1</sup>

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 9.

1 10<sup>12</sup> m What is the wavelength of light that has a frequency of 3 10<sup>4</sup> Hz in a vacuum? 10.

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3.3 10 19 J What is the energy of a photon that has a frequency ...

~~4 Arrangement of Electrons in Atoms~~

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Chemical Bonding

CHAPTER 6 REVIEW

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Chemical Bonding

SECTION 4 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

           In metals, the  
valence electrons are  
considered to be (a)  
attached to particular  
positive ions.

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~~Chapter 6 Section 4~~

~~Review Answers~~

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... By .... This is the  
compelling story of a  
damaged young woman,  
Nadia, who has taken  
refuge in a cabin in  
the....

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## CHAPTER 6 REVIEW

Chemical Bonding

### SECTION 6-2 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

Use the concept of

potential energy to

describe how a covalent

bond forms between two

atoms. As the atoms

involved in the

formation of a covalent

bond approach each

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Chapter 6

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Chapter Test Name

Class Date Chapter Test

A, continued Use this

figure to answer

questions 7 and 8.

\_\_\_\_\_ 7. A solution

containing 35 g of Li

$\text{2SO}_4$  dissolved in 100

g of water is heated

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from 10°C to 90°C.

According to information in the figure, this temperature

change would result in

a. an additional 5 g of Li<sub>2</sub>SO<sub>4</sub> in ...

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~~Test A Ed W. Clark~~

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Name Section Quiz,

continued Class Date 6.

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Metals are malleable because when struck, one plane of metal atoms can slide past another plane without breaking bonds. b. cannot easily move out of the way. c. moves in a way that maximizes the repulsive forces within the

~~Home David Brearley~~

~~High School~~

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Assignment Modern

Chemistry. CH 1

Vocabulary-New. CH 1

Mixed Questions. CH 1

Matter & Energy

Vocabulary-New ...

Notes: Chapter 6 CH 6

Chemical Bonding .

Reference: CH 6

Common Ion Table. ...

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6.1 & 6.2. Ch 6 Section

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Section Review 6.4.

Chapter 6

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Chapter 12 Review 2

Multiple Choice

Identify the letter of the choice that best

completes the statement or answers the question.

1. binds the atoms together is called a(n) A mutual electrical attraction between the nuclei and valence

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electrons of different  
atoms that a. dipole. c.  
chemical bond. b. Lewis  
structure. d. London  
force. 2. a.

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